

Instructions for completing CWM-TRK-05 Form

1. When completing the form please print legibly and complete ALL fields. Incomplete forms can result in delayed waste pickups.
2. Enter jerrican location.
3. Your University ID number is found on your University I-card (University ID). If you do not know what it is, you can look it up at:
https://www.icard.uillinois.edu/secure/act_tellme.cfm
4. Sign your name. This should be the same name printed for "Request submitted by" or the supervisor. The purpose of this signature is to satisfy legal requirements for identification of waste. By signing, you are stating that the attached information is correct, and saves the University from performing costly analysis of your waste.
5. Before submitting the form, enter jerrican number (unique number written on jerrican), pH of mixture, size of jerrican (10 or 20 L – 20L Jerrican should only be filled half full), and amount of solvent or oil in the jerrican.

As you add waste to the jerrican complete the information as shown in the example above.

1. Print the chemical name.
2. Print the amounts of the chemical added **each time** the chemical is added to the Jerrican.

Schedule — Drop off CWM-TRK-05 forms in the drop box located in the RAL nitrogen room. The deadline is Monday at 9:00 am for Tuesday pickup and Thursday at 9:00 am for Friday pickup.

Problem chemicals that **SHOULD NOT** be disposed in Jerricans.

This list contains chemicals that are known to commonly react when bulking solvents, create odor problems or are too toxic to bulk. Keeping these chemicals out of your jerrican will help the DRS be able to return your jerrican. DRS will be unable to return jerricans that have reactive chemicals. Chemicals known to react with solvent waste or otherwise extremely toxic should never be added to a jerrican. This list is not all-inclusive.

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|------------------------------------|---|--|--|
| 1. Allyl Alcohol | 14. Silicon and germanium hydroalkyls | 23. Chloroformates | 35. Sulfuric acid (conc.) |
| 2. Amines | 15. Zinc and cadmium alkyls | 24. Alpha cyanohydrins | These have a potential for polymerization if present in concentrations greater than 10%: |
| 3. Mercaptans | 16. Alkaline and alkaline earth hydrides and alkyls in solution | 25. Sulfite esters | 36. Vinyls |
| 4. Phosphines | 17. Lithium aluminum hydride | 26. Pyrocarbonate esters | 37. Nitriles |
| 5. Phosphite esters | 18. Sodium or calcium hydride | 27. Aziridine | 38. Carbonyls |
| 6. Isocyanides | 19. Iron pentacarbonyl | 28. Peroxides – not just the trace contamination of auto oxidation | 39. Ethers |
| 7. Alkynes | 20. Alkyl silyl halides | 29. Nitro esters | 40. Sulfones |
| 8. Dienes | 21. Acyl halides | 30. Nitroso esters | 41. Pyridines |
| 9. Thio ketones or esters | 22. Sulfonyl halides | 31. Hypochlorite esters | 42. Aromatics |
| 10. Carbon disulfide | | 32. Chromate esters | 43. Halogen 1,1 disubstituted vinyls |
| 11. Arsines | | 33. Hydrochloric Acid | |
| 12. Boranes | | 34. Nitric Acid | |
| 13. Aluminum and gallium trialkyls | | | |